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1. What is the mass of 7.50 moles of sulfur dioxide (SO_2)? TlFW dinitrogen tetroxide (N_2O_4)? ($\text{N} = 14.0 \text{ g/mol}$; $\text{O} = 16.0 \text{ g/mol}$)

2. What is the mass of 0.438 mole of ammonium chloride (NH_4Cl)? ~ carbon-oxygen compound, given that a 95.2-g sample is 43.0% carbon.

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